*Journal of Organometallic Chemistry,* 51 (1973) 153-166 .@ **Elsevie: Sequoia SA., Lausanne - Printed in The Netherlands-**

### DERIVATIVES OF DIVALENT GERMANIUM, TIN AND LEAD -.

# II\*. **SPECTROSCOPIC INVESTIGATION. OF- DItiCLOPENTADIEWL--**  TIN(II) AND ITS METHYLCYCLOPENTADIENYL ANALOGUE

**153:** 

:

. .

-

### P. G. HARRISON and M. A. HEALY

**Deparrmenr** *of Chemistry,* **Unicersity** *of Nottingham,* **University Park,** *Nottinghwn NG7 2RD (Great Britain)*  **(Received August 28th, 1972)** 

## **SUMMARY**

**The infra-red and Raman spectra of dicyclopentadienyltin(II) and its methylcyclopentadienyl analogue have been recorded and the observed bands assigned in terms of the "local" symmetry of the angular sandwich molecules. Proton magnetic**  resonance, mass and tin-119m Mössbauer spectra of bis(methylcyclopentadienyl)tin-**(II) are also reported\_** 

## **INTRODUCTION**

In sharp contrast to the vast preponderance of organotin(IV) compounds, **known organic derivatives of divalent tin, for which spectroscopic, i.e. tin-l 19m MGss**bauer isomer shift data, or structural confirmation is available, are as yet restricted to dicyclopentadienyltin<sup>2,3</sup> and its methylcyclopentadienyl analogue<sup>3</sup>, their boron trifluoride adducts<sup>4</sup>, monocyclopentadienyltin chloride and bromide<sup>5</sup>, bis(phenylbarenyl)tin<sup>6</sup>, and the unusual tin(II)-benzene complex,  $(C_6H_6)Sn(AlCl_4)_2 \cdot C_6H_6$ <sup>7</sup>. Moreover, the properties of tin(II)-carbon bonded compounds are vastly different to the rather inert  $\sigma$ -bonded skeleton<sup>1,8</sup> characteristic of the higher oxidation state.

**An electron diffraction study.has corroborated the earlier suggestion based on**   $\inf$ <sub>a-red</sub><sup>3.10.11</sup> and dipole moment<sup>12</sup> data that dicyclopentadienyltin possesses an angular sandwich structure in which the planar  $C<sub>5</sub>H<sub>5</sub>$  rings are centrally  $\sigma$ -bonded (pseudo  $\pi$ ) to the central metal atom, which is approximately  $sp^2$ -hybridised. The lone pair of electrons therefore occupies an orbital with directional character, and this has been illustrated by its use as a ligand in complex formation<sup>4</sup>. This mode of bonding may be compared with the localised  $\sigma$ -bonding of the puckered, diene-type  $C_5$  rings of tin(IV) cyclopentadienyls<sup>13,14</sup>. The infra-red spectra of dicyclopentadienyltin and -lead have been the subject of controversy<sup>3,10,11</sup>. In order to resolve this debate, we report in this paper the infra-red and Raman spectra of the former compound and its methylcyclopentadienyl analogue, together with related proton magnetic resonance, mass, and tin-119m Mössbauer spectral data.

**\*** For Part I see ref. 1.

. .

## **EXPERIMENTAL**

**All manipulations were carried out under an atmosphere of dry argon or nitro**gen. The organotin(II) compounds were prepared as described previously<sup>1</sup>, and su**blimed or distilcd** *in uacuo* **just before use. Infra-red spectra were recorded as nujol or**  halocarbon oil mulls, and as a carbon tetrachloride solution or liquid film using a Perkin-Elmer 521 spectrometer calibrated with polystyrene film and water vapour. Raman **spectra were recorded using a Cary 8.1 instrument equipped with a heliumneon laser. Proton magnetic resonance studies were performed using a Varian HA-100 instrument. Mass spectra were recorded at 70 eV by an AEI MS9 instrument.** 

# **RESULTS AND DISCUSSION**

**In this study we shall assume that the angular sandwich structure determined**  for dicyclopentadienyltin in the vapour phase<sup>9</sup> is preserved in both the solid and in **carbon tetrachloride solution. Also we shall assume that bis(methylcyclopentadienyl) tin possesses similar gross structural features. The highest possible overall symmetry**  for these molecules will therefore be  $C_{2v}$ .

Using the approximation of "local" symmetry the fundamental modes may be divided into two types: (i) ring vibrations, and *(ii)* vibrations associated with the tin **atom. We shall consider first the vibrations associated with the cyclopentadienyl and methylcyclopentadienyl rings each in turn, and later the skeletal modes of both molecules which involve the metal atom.** 

# *Ring modes of the cyclopentadienyl ring*

On the basis of  $C_{5v}$  "local" symmetry, the twenty-four normal vibrations of the **isolated C5H5 unit are distributed as** 

$$
3A_1 + A_2 + 4E_1 + 6E_2
$$

of which ten  $(2A_1 + A_2 + 3E_1 + 4E_2)$  will be in the plane of the ring and four  $(A_1 + E_1 +$ *2E,)* **perpendicular. The description of these modes is given in Table 1.** 

The observed infra-red and Raman bands are listed in Table 2. No gross **changes are apparent in going from the solid phase to a carbon tetrachloride solution. The assignment of the observed bands in this study are considerably aided by analogy**  with previous assignments of the spectra of  $\pi$ -cyclopentadienyl-metal carbonyl complexes<sup>15,16</sup> and ferrocene<sup>17</sup>, which have been confirmed by deuteration studies.

The five CH stretching modes  $[v_1(A_1), v_5(E_1), v_9(E_2)]$  are readily assigned to the two bands at ca.  $3100 \text{ cm}^{-1}$  in both the infra-red and the Raman. Polarisation studies of the similar two bands present in the Raman spectrum of  $\pi$ -cyclopentadienylmanganese tricarbonyl has unequivocally shown that the higher frequency band is the  $A_1$  mode in that case<sup>16</sup>. In the present case, the intensity of the higher frequency band in the Raman is substantially greater than the band at lower energy, and so is also assigned to the totally symmetric  $A_1$  mode,  $v_1$ .

The ring breathing mode  $v_3$ , also  $A_1$ , is easily identified as a strong band in both the infra-red and Raman at  $1116 \text{ cm}^{-1}$ . The remaining stretching vibrations to be assigned are the two CC stretching modes  $v_8$  of  $E_1$  symmetry, which should be active in both the infra-red and Raman, and  $v_{12}$  of  $E_2$  symmetry which should exhibit

**Raman activity only. The**  $E_1$  **mode occurs as a strong band at 1430 cm<sup>-1</sup> in both the infra-red and Raman. It would be tempting to assign the Raman band of medium'**  intensity at 1354 cm<sup>-1</sup> to the  $E_2$  mode. However, by analogy with previous assignments<sup>16,17</sup>, this vibration is expected to occur at higher energy than the  $E_1$  mode, and the weak band at  $1534 \text{ cm}^{-1}$  in the infra-red appears to be the most likely candidate.

Six CH bending modes are expected for  $C_{5v}$  symmetry. Of these, three will be in the plane of the ring  $[v_4(A_2), v_6(E_1), v_{10}(E_2)]$  and three perpendicular to the ring  $[v_2(A_1), v_7(E_1), v_{11}(E_2)]$ . Intense bands at 757 and 790 cm<sup>-1</sup> in the infra-red and similar weak bands in the Raman are immediately identified as  $v_2$  and  $v_7$  respectively, **although occurring to substantially lower energy of the same fimdamentals in cyclo**pentadienylmetal carbonyl complexes<sup>15,16</sup>. The third out-of-plane CH bending mode is expected to occur at  $\approx 1050$  cm<sup>-1</sup>, and is assigned to the weak infra-red and Raman active band at 1063 cm<sup>-1</sup>. Of the three in-plane modes, only the  $E_1$  mode ( $v_6$ ) is easily **recognised, and is assigned as a strong infra-red and weak Raman band** at ca. **1000**  cm<sup>-1</sup>. From previous studies<sup>15-17</sup> the two other in-plane CH bending modes are  $\texttt{expected in the regions:} \, A_2,1200-1250\,\texttt{cm}^{-1}, \, E_2 \approx\!1180\,\texttt{cm}^{-1}.$  These modes may not

## **TABLE 1**

### **DESCRIPTION OF RING MODES ON THE BASIS OF "LOCAL" SYMMETRY**



**u Becomes CMe out-of-plane bending mode. b Becomes CMe stretching mode. ' Becomes CMe in-plane bending mode.** 





s las

be assigned unequivocally, and strictly according to  $C_{5v}$  selection rules both should be inactive in the infra-red and only the  $E_2$  mode should exhibit Raman activity.<br>Weak bands at 1263 cm<sup>-1</sup> ( $v_4$ ,  $A_2$ ) and  $\approx$  1170 cm<sup>-1</sup> ( $v_{10}$ ,  $E_2$ ) appear to be the most probable choices.

The two remaining fundamentals to be assigned are  $v_{13}$  and  $v_{14}$ , the in-plane and out-of-plane CC bending vibrations respectively. Both these modes are of  $E_2$ symmetry, and so are strictly forbidden in the infra-red. They have previously been assigned in the ranges 900-950 cm<sup>-1</sup> and 480-620 cm<sup>-1</sup> respectively<sup>15-17</sup>. Weak infra-red bands at 914 and 960 cm<sup>-1</sup> may be assigned to the  $v_{13}$  mode, however they may equally well be combination bands. The band at ca. 890 cm<sup> $-1$ </sup>, which is also present in the Raman, is preferred for this mode, although this also may be accounted for as a combination. The out-of-plane mode is tentatively assigned to the weak band observed previously by Fritz<sup>10</sup>, but not by us, at 544 cm<sup>-1</sup>. This band was assigned by Fritz as the antisymmetric metal-ring tilt mode, but this fundamental is to be expected at much lower energy<sup>16</sup>. A summary of the assignments for  $(C_5H_5)_2$ Sn together with those for related compounds is given in Table 3.

#### TABLE 3

SUMMARY OF THE ASSIGNMENTS OF THE RING VIBRATIONS OF  $(C_5H_5)_2$ Sn AND RELATED **COMPOUNDS** 



"This work. <sup>b</sup>Ref. 16. 'Ref. 18. <sup>d</sup>Ref. 17.

Rina modes of the methylcyclopentadienyl rina

Although the isolated methylcyclopentadienyl ring is nominally of the  $C$ . molecular point group, if the methyl group is considered as a point mass, the symmetry is raised to  $C_{2v}$ . On this basis, the fourteen normal modes of the cyclopentadienyl ring become twenty-four normal modes, distributed among the symmetry species as:

$$
9A_1 + 3A_2 + 8B_1 + 4B_2
$$

as the degeneracy of the  $E_1$  and  $E_2$  modes under  $C_{5v}$  symmetry is removed. The description of the twenty-four normal modes is given in Table 1, which also indicates

P. G. HARRISON, M. A. HEALY

their origin to the fundamentals of the unsubstituted ring. The modes  $v_{7b}$ ,  $v_{9a}$  and  $v_{10b}$  then become the out-of-plane and in-plane CMe bending and CMe stretching vibrations respectively. In addition, there will be nine internal modes of vibration of the methyl group. These are described in Table 4.

### TABLE 4

DESCRIPTION OF INTERNAL VIBRATIONS OF THE METHYL GROUP OF THE MeC<sub>5</sub>H<sub>4</sub> **RING** 



"Frequency number.

The observed infra-red and Raman bands together with polarisation data are listed in Table 5. The most profitable approach to the assignment of the fundamental modes of the MeC<sub>5</sub>H<sub>4</sub> rings of bis(methylcyclopentadienyl)tin is to use the assignments for the unsubstituted ring, utilizing the inequality rule of Whiffen and Steele<sup>19</sup>. The assignment of the vibrational spectra of dimethylferrocene<sup>20</sup> and some (methylcyclopentadienyl)metal<sup>21</sup> carbonyl complexes have been assigned in a similar way. It is worthwhile mentioning here that, since no vibration of the rigid methyl group is attributable to an  $A_2$  species, the inequality rule may not be applied to this species, and the observed frequency for  $(MeC_5H_4)_2$ Sn should resemble corresponding frequencies in  $(C_5H_5)_2Sn$ .

First of all we shall consider vibrations associated with the methyl group, described in Table 4. The three CH stretching modes  $(A_1 + B_1 + B_2)$  may be immediately assigned to the two bands in the infra-red at 2965 and 2925  $cm^{-1}$ . The latter band occurs as a polarised band in the Raman and so is assigned as the  $A_1$  mode,  $m_1$ . Parker and Stiddard<sup>21</sup> have similarly assigned the lower frequency band to the  $A_1$  mode by analogy with dimethylferrocene and toluene. The three HCH bending modes  $(A_1 +$  $B_1 + B_2$ ) should occur in the region 1370–1500 cm<sup>-1</sup>. A suitable candidate for the  $A_1$ mode is the weak band at ca. 1380 cm<sup>-1</sup> which is polarised in the Raman. The  $B_1$  and  $B_2$  modes,  $m_5$  and  $m_8$ , are to be expected at higher frequency. In dimethylferrocene and MeC<sub>5</sub>H<sub>4</sub>Mn(CO)<sub>3</sub> these modes have been assigned to bands at ca. 1460 cm<sup>-1</sup> and ca. 1480 cm<sup>-1</sup> respectively. Of the bands available in this region, the infra-red and depolarised Raman band at 1482 cm<sup>-1</sup> is assigned to the  $B_2$  mode,  $m_8$ , on account of its greater intensity than the weak shoulder at 1492 cm<sup>-1</sup>. The choice for the  $B_1$ mode lies between weak bands at  $1458 \text{ cm}^{-1}$  and  $1448 \text{ cm}^{-1}$ . The latter may be assigned to the higher frequency band. Three methyl bands remain to be assigned: the inplane and out-of-plane CMe wagging modes,  $m_6(B_1)$  and  $m_9(B_2)$ , and the Me torsion

158

# **TABLE 5**

۰.

 $\bar{\gamma}$ 

98



**OBSERVED VIBRATIONAL SPECTRA OF BIS(METH** 

**0** In cm<sup>-1</sup>; w = weak, mw = medium weak, = medium, ms = medium strong, s = strong, v = very,  $\sin\theta$  =  $\mathbf{s}$ houlder, (br) = broad, p = polarised, dp = depolarise I

mode  $(A_2)$ . This latter mode, strictly inactive in the infra-red for  $C_{2v}$  symmetry, is expected to occur at a frequency below  $250 \text{ cm}^{-1}$ , outside the range of this study, **although the Raman band at 113 cm- ' cannot be ruled out. Bands at 1062 w, 1042 m,**  1026 m and 1010 w(sh) cm<sup>-1</sup> in the infra-red may be considered for  $m_6$  and  $m_9$ . A single band at 1046 cm<sup>- 1</sup> has been assigned to both modes in  $\mathrm{MeC}_5\mathrm{H}_4\mathrm{Mn}(\mathrm{CO})_3$ whilst for dimethylferrocene<sup>20</sup> bands at 1037 cm<sup>-1</sup> and 1053 cm<sup>-1</sup> have been assigned as the  $B_1$  and  $B_2$  modes respectively. The band at 1043 cm<sup>-1</sup> is polarized in the Raman and hence is assigned to the ring breathing mode  $(A<sub>1</sub>)$   $v<sub>3</sub>$ . The very weak shoulder at  $1010 \text{ cm}^{-1}$  is probably too low for a wagging mode, whilst the weak infra-red and depolarised Raman band at 1062 cm<sup>-1</sup> is probably the ring in-plane CH bend,  $v_{6b}$ . **The CMe wagging modes are therefore assigned to the medium intensity band at**   $1026 \text{ cm}^{-1}$ .

Three of the ring modes,  $v_{7b}$ ,  $v_{9a}$  and  $v_{10b}$ , are associated with the rigid methyl group. The CMe stretch,  $v_{9a}(A_1)$ , is assigned to the strong band at 1229 cm<sup>-1</sup>, which **also occurs as a strong polarised Raman line, consistent with previous assignments' '. The band at ca. 320 cm-' in both the infra-red and Raman, absent in the spectra of**   $(C_5H_5)$ <sub>2</sub>Sn has similarly been assigned to the in-plane CMe bending mode,  $v_{10b}$ . The out-of-plane bend,  $v_{7b}$ , has been assigned to a band at 217 cm<sup>-1</sup> in toluene<sup>21</sup>, and is **not assigned in this present work.** 

**The remaining ring fundamentals may be assigned using the assignments for**   $(C_5H_5)_2$ Sn and utilizing the inequality rule. The ring CH stretching modes,  $2A_1 + 2B_1$  $(v_1, v_{5a}, v_{5b}, v_{9b})$  are assigned to the two bands observed in both the infra-red and Raman at 3080 and 3100 cm<sup>-1</sup>. The assignment by previous authors<sup>21</sup> of the higher frequency band to the  $A_1$  modes is confirmed in the present case by the polarisation **properties of the bands.** 

As stated above, the ring breathing mode,  $v<sub>3</sub>$ , is readily assigned to the infra-red and polarised Raman band at  $1042 \text{ cm}^{-1}$ . This is significantly lower than that in the **unsubstituted ring, indicating appreciable interaction** between this mode and **the**  CMe stretching vibration. Two other CC stretching vibrations,  $v_8$  and  $v_{12}$ , need to be assigned. These modes, originally  $E_1$  and  $E_2$  modes respectively in  $(C_5H_5)_2\text{Sn}$ , become split into  $A_1$  and  $B_1$  compounds under  $C_{2v}$  symmetry. For  $v_8$ , the inequality rule predicts the range  $1534-1429$  cm<sup>-1</sup> for the  $A_1$  species, and  $1429-1263$  cm<sup>-1</sup> for the  $\overline{B}_1$  species. Of the band of the higher range, only the weak features at 1492 (sh) and **1448 cm- ' remain unassigned. In the lower range, a band at 1408 cm- ', which is also present as a depolarised Raman line, seems to be the obvious choice for the B, mode.**  This would then make the band at  $1448 \text{ cm}^{-1}$  the most suitable candidate for the  $A_1$ mode. The  $v_{12}$  mode was assigned to the band at 1534 cm<sup>-1</sup> in  $(C_5H_5)_2$ Sn. The inequality rule therefore predicts the ranges 3090-1534 and 1534-1429  $cm^{-1}$  for the  $A_1$ and *B*<sub>1</sub> respectively. Bands at 1609 and 1558 cm<sup>-1</sup> have been assigned to these modes in dimethylferrocene<sup>20</sup>, whilst in MeC<sub>5</sub>H<sub>4</sub>Mn(CO)<sub>3</sub><sup>21</sup>, bands attributable to these **modes have not been observed but have been estimated to occur at 1600 and 1500**   $cm^{-1}$  respectively. Of the several bands available in the spectrum of  $(Mec_5H_4)_2Sn$ , those at 1607 and 1526 cm<sup>-1</sup> have been assigned to the  $A_1$  and  $B_1$  modes by comparison **with dimethyIferrocene20.** 

**The four in-plane CH bending modes belong to the symmetry species 2A,-**   $(v_{6a}, v_{10a}) + 2B_1(v_4, v_{6b})$ . Application of the inequality rule would place the  $A_1$  modes,  $v_{10a}$  and  $v_{6a}$ , in the ranges 1161–1114 cm<sup>-1</sup> and 1011–886 cm<sup>-1</sup> respectively. In these **ranges, only two polarised Raman bands, 1114 and 980 cm-', have not yet been**  assigned, the polarised band at  $1043 \text{ cm}^{-1}$  having been assigned to the ring breathing mode  $v_3$ . These bands are therefore assigned to the modes  $v_{10a}$  and  $v_{6a}$  respectively. The  $B_1$  modes are, however, not so easily assigned. The inequality rule predicts the ranges 1161-1011 cm<sup>-1</sup> for  $v_{6b}$  and 1263-1161 cm<sup>-1</sup> for  $v_4$ . In the former range, only the weak infra-red band at 1062 cm<sup>-1</sup>, depolarised in the Raman, is unassigned, and so is assigned as  $v_{6b}$ . Bands in the latter range occur at 1260, 1236, 1229 and 1163 cm<sup>-1</sup>. The band at 1229 cm<sup>-1</sup> has already been assigned as the CMe stretching mode  $v_{9a}$ . The band at  $1260 \text{ cm}^{-1}$  also occurs as a depolarised line in the Raman, and so is the preferred assignment for  $v_A$ .

**The remaining in-plane vibration to be assigned is the CC in-plane bending**  mode,  $v_{13}$ . In  $C_{2v}$  symmetry, this mode becomes split into  $A_1(v_{13a})$  and  $B_1(v_{13b})$ components, for which the inequality rule predicts that  $v_{13a}$  should fall below 886  $cm^{-1}$  and  $v_{13b}$  should lie in the range 1011–886 cm<sup>-1</sup>. Bands at 851 cm<sup>-1</sup> (A<sub>1</sub>) and **892 cm- l (B,) have been assigned to these modes in dimethylferrocene; corresponding**  to the 885 cm<sup>-1</sup> band in ferrocene itself<sup>20</sup>. In the present case, the choice appears to **lie between the three weak features in the infra-red at 847,885 and 898 cm- i, the low**  frequency band being assigned as  $v_{13a}$ .

**Bending vibrations perpendicular to the ring belong to the symmetry species**   $A_2$  and  $B_2$ . As noted above, the inequality rule does not apply for the  $A_2$  modes, and **those fundamentals should closely resemble those in the unsubstituted ring. Thus the**  out-of-plane CH bending mode  $v_{7a}$  is readily located as a strong infra-red and weak **Raman band at 813 cm<sup>-1</sup>. The**  $A_2$  **modes should strictly be inactive in the infra-red and the intensity of this band may indicate some lowering of the symmetry of the ring**  from  $C_{2n}$ . The other out-of-plane CH bending mode of  $A_2$  symmetry  $v_{11a}$  should  $\text{occur}$  at ca. 1062 cm<sup>-1</sup>, the position of  $v_{11}$  in  $\text{(C}_5H_5)_2$ Sn. The weak infra-red and Raman band at 1062 cm<sup>-1</sup> in the spectra of  $(\text{MeC}_5H_4)_2\text{Sn}$  has already been assigned as the in-plane mode  $v_{6b}$ , but it is conceivable that both modes occur at this frequency. However in MeC<sub>5</sub>H<sub>4</sub>NiNO<sup>22</sup> and dimethylferrocene<sup>20</sup>, this mode has been assigned to bands at 967 and 1002 cm<sup>-1</sup> respectively, and so the weak feature at  $1010 \text{ cm}^{-1}$  is **not unreasonable in the present case. Of the two remaining out-of-plane CH bending**  modes, both  $B_2$ , the mode  $v_2$  is immediately assigned as the intense infra-red band at 767 cm<sup>-1</sup>, whilst the inequality rule places the position of  $v_{11b}$  in the range 790-1062 **cm-'** . **In this range, only weak features at 976,939(sh) cm- ' and one of the two bands**  at 898 and 885 cm<sup>-1</sup> (the other already having been assigned to  $v_{13b}$ ) remain unassign**ed\_ In several (methylcyclopentadienyl)metal carbonyl complexes, this mode was**  thought<sup>21</sup> to occur in the range  $989-982$  cm<sup>-1</sup>. In the present case, the assignment of  $v_{11b}$  is at best tentative, the best candidate probably being the band at 976 cm<sup>-1</sup>.

**Finally, the out-of-p!ane CC bending modes need to be assigned. This mode,**   $v_{14}$ , was tentatively assigned to a band at 544 cm<sup>-1</sup> in  $(C_5H_5)_2$ Sn. In  $(Mec_5H_4)_2$ Sn this mode becomes split into two components of the symmetry species  $A_2$  and  $B_2$ . In dimethylferrocene these have been assigned to 597 and 632 cm<sup>-1</sup> respectively, whilst in toluene they occur at 408 and 464 cm<sup>-121</sup>. The  $A_2$  mode should occur near 544 cm<sup>-1</sup>, and is assigned as the weak infra-red and Raman band at 561 cm<sup>-1</sup>. The choice for the  $B_2$  mode lies between the infra-red bands at 585,610 and 630 cm<sup>-1</sup>, the **two latter bands also occurring in the Raman. The 630 cm-' band is probably too**  high to be associated with a band at 544 cm<sup>-1</sup>, and so  $v_{14b}$  is assigned to the band at

### TABLE 6

SUMMARY OF THE ASSIGNMENTS OF THE RING VIBRATIONS OF (MeC<sub>5</sub>H<sub>4</sub>)<sub>2</sub>Sn AND RELATED COMPOUNDS

Species	$(MeC5H4)2Sn$	MeC <sub>5</sub> H <sub>4</sub> NiNO <sup>2</sup>	$(MeC_5H_4)_2Fe^b$
felt w $A_{1}$	3100	3090	3087
	3100	3090	3087
	1607	1610	1609
	1448	1452	1440
	1229	1225	1229
	1117		1106
	1042	1042	1025
	928	928	919
	847	849	851
v(Me)	2925	2911	2922
$\delta(Me)$	1380	1385	1384
$B_{1}$	3080	3106	3087
	3080	3106	3087
	1526	(1560)	1558
	1408	1385	1373
	1260	1237	1207
	1062	1025	1115
	898 or 885	888	892
	319	344	328
v(Me)	2965	2941	2950
$\delta$ (Me)	1458	1460	1464
$\delta$ (Me)	$-1026$	1056	1037
$\boldsymbol{B}_2$	976	975	975
	767	802	811
	610	624	632
		354	(217)
vMe	2965	2973	2966
$\delta M$ e	1482	1481	1478
$\delta M e$	1026	1056	1053
$A_{2}$	1010	967	1002
	813	819	833
	561	780 or 639	597

" Ref. 22. " Ref. 20.

610 cm<sup>-1</sup>. A summary of the assignments of  $(MeC_5H_4)_2Sn$  is given in Table 6, together with those of the related molecules MeC<sub>5</sub>H<sub>4</sub>NiNO<sup>22</sup> and dimethylferrocene<sup>20</sup>.

# Non-ring skeletal modes

Assuming the C<sub>5</sub> rings to be rigid, nine skeletal modes  $(3A_1 + 2A_2 + 2B_1 + 2B_2)$ involving the tin atom will be expected for angular dicyclopentadienylmetal compounds under  $C_{2n}$  symmetry. The description of these modes is given in Table 7.

Bands attributable to the ring tilting modes  $(2A_1 + B_1 + B_2)$  have not been observed in  $(C_5H_5)_2$ Sn. However, in  $(Mec_5H_4)_2$ Sn, the bands at 368 vvw and 295 m(sh)  $cm^{-1}$  in the infra-red could belong to such vibrations. In the analogous ferrocenes, the

 $162 -$ 

## **TABLE7.**

# **DESCRIPTION OF THE NON-RING SKELETAL MODES FOR ANGULAR DICYCLOPENTADI-ENYLMETAL COMPOUNDS**

Under C<sub>2</sub>, symmetry.



symmetric and antisymmetric modes have been reported to occur at ca. 380 cm<sup>-1</sup> and ca. 495 cm<sup>-1</sup> respectively; and  $C_5H_5NiNO$  and  $MeC_5H_4NiNO$  the ring tilting mode **occurs at 252 and 298 cm-' respectively.** 

In the Raman spectra of both compounds two bands at ca. 240 s cm<sup>-1</sup> and ca. **170 w-m cm-l were observed. The higher frequency band was also observed in the**  infra-red spectra, but an accurate measurement was not possible, since it occurred on the very limit of this study. Polarisation studies for  $(MeC<sub>5</sub>H<sub>4</sub>)<sub>2</sub>Sn$  show that the band at 171 cm<sup>-1</sup> belongs to the  $A_1$  symmetry class, whilst the 240 cm<sup>-1</sup> band is depolarised. **These bands are therefore assigned as the antisymmetric and symmetric metal-ring**  stretching modes,  $R_2$  and  $R_1$ . These assignments are much lower than the similar **assignments for analogous sandwich molecules (Table 8), but are quite reasonable in view of the greater mass of the tin atom. Fritz, who observed a strong band in the**  infra-red of  $(C_5H_5)_2$ Sn at 260 cm<sup>-1</sup>, presumably the same as our 242 cm<sup>-1</sup> Raman band, has compared the tin- $C_5H_5$  ring stretching mode with the tin-bromine stretch, **since both ligands are of comparable mass". This vibration may be found in the range**  170-280 cm<sup>-1</sup> in tin(IV) complexes<sup>23-26</sup>, and in the range 182-205 cm<sup>-1</sup> in tin(II) derivatives<sup>27</sup>. The metal–ring stretching vibration may also be likened to the Whiffen *t*-mode in phenyltin compounds. This mode describes what is for the most part the **Sn-Ph stretching vibration, and usually occurs as a strong band in the infra-red26 at ca. 220 cm-'.** 

# **FABLE 8**

### **iKELETAL MODES IN SOME "SANDWICH" MOLECULES**



Ref. 17. <sup>b</sup> Ref. 16. Ref. 28. <sup>d</sup> Ref. 18.

The medium intensity depolarised line at 113 cm<sup> $-3$ </sup> in the Raman spectrum of  $(MeC<sub>5</sub>H<sub>4</sub>)<sub>2</sub>$ Sn could be assigned to either of the torsion modes  $R<sub>2</sub>$  or  $R<sub>4</sub>$ , or the ringmetal-ring deformation mode  $R<sub>9</sub>$ . Although the torsion modes cannot be ruled out, the deformation mode  $R_9$  is preferred. The corresponding vibration in ferrocene is found at 170 cm<sup>-1</sup>, whilst the similar deformation modes in  $C_5H_5Mn(CO)_3$ ,  $(C_6H_6)$ , Cr and  $C_5H_5NiNO$  fall in the range 100–150 cm<sup>-1</sup>.

It is clear that the vibrational spectra of  $(C_5H_5)_2$ Sn and  $(MeC_5H_4)_2$ Sn are quite adequately described in terms of the "local" symmetries for the angular sandwich molecules.

Further information concerning their structural similarity comes from the dired ai susteel tasaiaara teoan solT .(R eldeX} ebauoayaas ovit edt lo extrege eesaa spectra is the polyisotopic monoring-tin fragment, with only a weak parent ion. Monocyclopentadienyltin chloride is reported to behave similarly<sup>5</sup>. Ferrocene on the other hand exhibits the parent ion as the strongest peak. Fragments analogous to the cyclopropeniumtin and alkynyltin species observed for  $(C<sub>5</sub>H<sub>5</sub>)$ , Sn could not be detected in the methylcyclopentadienyl compound.

#### TABLE 9

MASSSPECTRA OF DICYCLOPENTADIENYLTINAND BIS(METHYLCYCLOPENTADIENYL)-**TIN** 



 $2$  Ref.  $3$ .

#### TABLE 10

TIN-119m MÖSSBAUER DATA FOR ORGANOTIN(II) DERIVATIVES



<sup>a</sup> Ref. 8. <sup>b</sup> Ref. 30. <sup>c</sup> Ref. 4.

The tin-119m Mössbauer spectral data of both compounds and  $(C_5H_5)$ , Sn·  $BF<sub>3</sub>$  is listed in Table 10. The isomer shift and quadrupole splitting parameters are all identical within experimental error, indicating that the valence electrons experience no significant perturbation as the groups about the tin atom are varied.

The proton magnetic resonance spectrum of  $(MeC<sub>5</sub>H<sub>4</sub>)<sub>2</sub>$  Sn at ambient tempera-



Fig. 1. Proton magnetic resonance spectrum of  $(MeC_5H_4)_2$ Su in the regions  $\tau$  4.30 and  $\tau$  8.0.

tures exhibits two resonances at ca.  $\tau$  4.30 and  $\tau$  7.97 (Fig. 1) [cf. Me<sub>3</sub>SnC<sub>5</sub>H<sub>4</sub>Me:  $\tau$  4.34, 4.47 (C<sub>5</sub>H<sub>4</sub>);  $\tau$  7.96 (Cp–CH<sub>3</sub>)]<sup>29</sup>. The low field resonance consists of a partially resolved AA'BB' pattern for the ring protons, whilst the resonance due to the methyl group consists of a sharp single line. Spin-spin coupling between the metal atom and the methyl protons was not observed, however partially resolved tin satellites could be seen on the low field resonance with  $J(Sn-H) \approx 10$  Hz [cf.  $(C_sH_s)_2Sn$ , J 159 Hz]<sup>8</sup>. This is to be compared with  $Me<sub>3</sub>SnC<sub>5</sub>H<sub>4</sub>Me$  where coupling was observed between tin and the methyl protons  $(J \t10.5 \text{ Hz})$  but not the ring protons<sup>29</sup>.

#### **REFERENCES**

- 1 P. G. Harrison and S. R. Stobart, J. Chem. Soc., Dalton Trans., (1973) in press.
- 2 E. O. Fischer and H. Grubert, Z. Naturforsch. B. 11 (1956) 423.
- 3 L. D. Dave, D. F. Evans and G. Wilkinson, J. Chem. Soc., (1959) 3684.
- 4 P. G. Harrison and J. J. Zuckerman, J. Amer. Chem. Soc., 92 (1970) 2577.
- 5 K. D. Bos, E. J. Bulten and J. G. Noltes, J. Organometal. Chem., 39 (1972) C52.
- 6 Yu. A. Aleksandrov, V. I. Bregadse, V. I. Goldanskii, L. J. Zakhardin, O. Yu. Okhlobystin and V. V. Kharpov, Dokl. Akad. Nauk SSSR, 165 (1965) 593.
- 7 H. Lüth and E. L. Amma, J. Amer. Chem. Soc., 90 (1968) 5941; 91 (1969) 7516.
- 8 P. G. Harrison and J. J. Zuckerman, J. Amer. Chem. Soc., 91 (1969) 6885.
- 9 A. Almenningen, A. Haaland and T. Motzfeldt, J. Organometal. Chem., 7 (1967) 97.
- 10 H. P. Fritz, Chem. Ber., 90 (1959) 780.
- 11 H. P. Fritz and E. O. Fischer, J. Chem. Soc., (1961) 547.
- 12 E. Weiss, Z. Anorg. Allg. Chem., 287 (1956) 236.
- 13 B. P. Biryukov, Yu. T. Struchkov, K. N. Anisimov, N. E. Kolobova and V. V. Skripkin, Chem. Commun.. (1968) 119; Zh. Strukt. Khim., 10 (1969) 95.
- 14 N. N. Veniaminov, Yu. A. Ustynyuk, N. V. Alekseev, I. A. Ronova and Y. T. Struchkov, Zh. Strukt. Khim., 12 (1971) 952.
- 15 D. J. Parker and M. H. B. Stiddard, J. Chem. Soc. A. (1970) 480.
- 16 I. J. Hyams, R. T. Bailey and E. R. Lippincott, Spectrochim. Acta, Part A, 23 (1967) 273.
- 17 E. R. Lippincott and R. D. Nelson, Spectrochim. Acta, 10 (1958) 307.
- 18 R. D. Feltham and W. G. Fateley, Spectrochim. Acta, 20 (1964) 1081.
- 19 D. Steele and D. H. Whilfen, Trans. Faraday Soc., 55 (1959) 369.
- 20 R. T. Bailey and E. R. Lippincott, Spectrochim. Acta, 21 (1965) 389.
- 21 D. J. Parker and M. H. B. Stiddard, J. Chem. Soc. A, (1970) 1040.
- 22 R. T. Bailey, Spectrochim. Acta, Part A, 25 (1969) 1127.

166 G. HARRISON, M. A. HEALY

23 C. W. Hobbs and R. S. Tobias, Inorg. Chem., 9 (1970) 1037.

24 R. J. H. Clark, L. Maresca and R. J. Puddephatt, Inorg. Chem., 7 (1968) 1603.

25 R. J. H. Clark, A. G. Davies and R. J. Puddephatt, J. Chem. Soc. A, (1968) 1828.

26 M. K. Das, J. Buckle and P. G. Harrison, Inorg. Chim. Acta, 6 (1972) 17.

27 R. J. H. Clark, L. Maresca and P. J. Smith, J. Chem. Soc. A, (1970) 2687.

28 H. P. Fritz, W. Lüttke, H. Stammreich and R. Forneris, Spectrochim. Acta, 17 (1961) 1068.

29 A. Davison and P. E. Rakita, J. Amer. Chem. Soc., 90 (1968) 4479; Inorg. Chem., 9 (1970) 289.

30 P. G. Harrison and J. J. Zuckerman, unpublished results.